

GROZIN, B.D., otv.red.; DRAYGOR, D.A., zam.otv.red.; SAMOKHVALOV, Ya.A., red.toma; BRAUN, M.P., red.; FAYNERMAN, I.D., red.; KRAGEL'SKIY, I.V., red.; BARABASH, M.L., red.: Prinimali uchastiye: VAYNBERG, D.V., prof.; PETRENKO, I.P., kand.tekhn.nauk; SINYAVSKAYA, M.D., inzh.; SHEVCHUK, V.A., kand.tekhn.nauk; SEMIROG-ORLIK, V.N., kand.tekhn.nauk; YANKOVICH, V.F., inzh.; GORB, M.L., kand.tekhn.nauk; RAKHLINA, N.P., tekhn.red.

[Increasing the wear-resistance and life of machinery] Povyshenie iznosostoikosti i sroka sluzhby mashin. Kiev, Izd-vo Akad.nauk USSR. Vol.2. 1960. 290 p. (MIRA 14:1)

1. Vsesoyuznoye nauchno-tekhnicheskoye obshchestvo mashinostroi-  
tel'noy promyshlennosti. Kiyevskoye oblastnoye pravleniye.  
(Mechanical wear) (Machinery)

SHEVCHUK, V. G.

Physical Chemistry

Dissertation: "Study of Quaternary Systems Consisting of the Sulfates of Sodium, Ammonium, Zinc, and Water by Methods of Physicochemical Analysis." Cand Chem Sci, Inst of General and Inorganic Chemistry imeni N. S. Kurnakov, Acad Sci USSR, Oct-Dec 1953. (Vestnik Akademii Nauk, Mar 54)

SO: SUM 213, 20 Sept 1954

# USSR .

Solubility diagram for the ternary system  $(\text{NH}_4)_2\text{SO}_4$ - $\text{ZnSO}_4$ - $\text{H}_2\text{O}$  at  $35^\circ$ . A. S. Karmukhov and V. G. Shevchuk (K. D. Vishinski State Pedagog. Inst., Yaroslavl). *Doklady Akad. Nauk S.S.S.R.* 90, 101-4(1953).—The ternary system  $(\text{NH}_4)_2\text{SO}_4$ - $\text{ZnSO}_4$ - $\text{H}_2\text{O}$  was studied at  $35^\circ$  by the isothermal method (cf. C.A. 48, 7414d). Equil. was reached after 18-20 hrs. after which time samples were taken for analysis. The soly. diagram of this system has 3 branches. The binary compd.  $\text{ZnSO}_4 \cdot (\text{NH}_4)_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$  was isolated and its microstructure and  $n$  were detd.

J. Rovtar Leach

SHEVCHUK, V.G.  
SHEVCHUK, V.G.

Physicochemical analysis of the ternary system  $\text{Na}_2\text{SO}_4$  --  $(\text{NH}_4)_2\text{SO}_4$  --  
--  $\text{H}_2\text{O}$ . Soob.o nauch.rab.chl.VKHO no.2:42-45 '54. (MIRA 10:10)  
(Sodium sulfates) (Ammonium sulfate) (Water)

SHEVCHUK, V.G.; LEPESHKOV, I.N.

Solubility isotherm of the quaternary system:  $\text{Na}_2\text{SO}_4$  -  $\text{ZnSO}_4$  -  
 $(\text{NH}_4)_2$  -  $\text{H}_2\text{O}$  at  $35^\circ$ . Zhur.neorg.khim. 1 no.8:1888-1895 Ag '56.  
(Sulfates) (MLRA 9:11)

5(4)

AUTHORS:

TITLE:

PERIODICAL:

ABSTRACT:

SOV/153-58-2-5/30  
 Shevchuk, V. G., Druzhinin, I. G.  
 The Solubility of Salts in the System of  $\text{ZnSO}_4 - (\text{NH}_4)_2\text{SO}_4 - \text{H}_2\text{O}$   
 at 35 and 50°  
 (Rastvorimost' soley v sisteme  $\text{ZnSO}_4 - (\text{NH}_4)_2\text{SO}_4 - \text{H}_2\text{O}$  pri 35 i 50°)

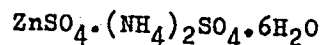
Izvestiya vysshikh uchebnykh zavedeniy. Khimiya i khimicheskaya tekhnologiya, 1958, Nr 2, pp 25-30 (USSR)  
 Only few data on the common solubility of zinc and ammonium sulfates in water are available as yet (Refs 1-3). The study of this system within a wide temperature range is necessary for establishing the production of pure zinc sulfate from mother lyes of some waste matters of the halurgical industry. A comprehensive survey of the corresponding literature written since M. V. Lomonosov's time is given (Refs 4-7). The authors also give a study of the ternary system, as mentioned in the title. Then an experimental section follows. Table 1 and figure 1 show that the solubility isotherm of zinc and ammonium sulfate in water at 35° is characterized by three branches.- By adding small quantities of ammonium sulfate a double salt can be immediately obtained. For that reason, the authors could never

Card 1/3

The Solubility of Salts in the System of  
 $\text{ZnSO}_4-(\text{NH}_4)_2\text{SO}_4-\text{H}_2\text{O}$  at 35 and 50°

SOV/153-58-2-5/30

obtain even one single element of the remainder of the curve branch concerned. Point 2 is eutonic. The branch from 3 to 16 corresponds to the crystallization of the compound:



Point 17 is eutonic too. Figure 2 and table 2 show that the temperature increase from 35° to 50° is favorable for the formation of double salt. The first branch of the curve from point 1 to 3 corresponds to the separation of  $\text{ZnSO}_4 \cdot 6\text{H}_2\text{O}$ . The branch of the double salt is somewhere between point 5 and 17. Points 5 and 18 are eutonic. By adding minimal quantities of zinc sulfate to the saturated solution of ammonium sulfate double salt is immediately obtained. In order to give a more complete characterization of the ternary system its solid phases were examined by physico-chemical analysis. The results obtained confirm as a whole the documentary data mentioned in references (Fig 3)(Table3). Moreover, the thermogram of the double salt concerned has been also deciphered. (Fig 4).

Card 2/3

The Solubility of Salts in the System of  
 $\text{ZnSO}_4-(\text{NH}_4)_2\text{SO}_4-\text{H}_2\text{O}$  at 35 and 50°

SOV/153-58-2-5/30

There are 4 figures, 3 tables, and 12 references, 7 of which are Soviet.

ASSOCIATION: Yaroslavskiy pedagogicheskiy institut  
(Yaroslav Institute of Pedagogy).  
Kafedra khimii  
(Chair of Chemistry)

SUBMITTED: September 10, 1957

Card 3/3



SHEVCHUK, V.G.

Equilibrium in the systems  $\text{Li}_2\text{SO}_4 - \text{MgSO}_4 - \text{H}_2\text{O}$ ,  $\text{Rb}_2\text{SO}_4 - \text{MgSO}_4 - \text{H}_2\text{O}$  at  $35^\circ$ . Zhur.neorg.khim. 6 no.8:1955-1958 Ag '61. (MLRA 14:8)

1. Poltavskiy institut inzhenerov sel'skokhozyaystvennogo stroitel'stva.  
(Systems (Chemistry))

SHEVCHUK, V.G.

Solubility in the system  $\text{Li}_2\text{SO}_4 - \text{Rb}_2\text{SO}_4 - \text{H}_2\text{O}$  at  $35^\circ\text{C}$ .  
Zhur.neorg.khim. 7 no.10:2463-2465 0'62. (MIRA 15:10)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra khimii.  
(Alkali metal sulfates) (Solubility)

SHEVCHUK, V.G.; KOST', L.L.

Equilibria in the systems  $\text{Cs}_2\text{SO}_4 - \text{MgSO}_4 - \text{H}_2\text{O}$  and  
 $(\text{NH}_4)_2\text{SO}_4 - \text{MgSO}_4 - \text{H}_2\text{O}$  at  $35^\circ$ . Zhur. Neorg. khim. 9 no.2:  
432-436 F'64. (MIRA 17:2)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra khimii.

SHEVCHUK, V.G.; KOST', L.L.

System  $\text{Li}_2\text{SO}_4 - \text{K}_2\text{SO}_4 - \text{MgSO}_4 - \text{H}_2\text{O}$  at  $35^\circ\text{C}$ . Zhur. neorg.  
khim. 9 no.5:1242-1245 My '64. (MIRA 17:9)

1. Poltavskiy inzhenerno-stroitel'nyy institut.

SHEVCHUK, V.G.; AVERINA, R.A.

System lithium sulfate-beryllium sulfate-magnesium sulfate-water  
at 35 degrees Centigrade. Zhur. neorg. khim. 9 no.12:2764-2768  
D '64. (MIRA 18:2)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra khimii.

SHEVCHUK, V.G.; VAYSFEL'D, M.I.

The system lithium chloride plus magnesium sulfate forms and is formed from lithium sulfate plus magnesium chloride at 35 degrees Centigrade. Zhur. neorg. khim. 9 no.12:2769-2774 D '64.  
(MIRA 18:2)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra khimii.

SHEVCHUK, V.G.

Some characteristics of the effect of adrenaline and insulin  
on the sugar level in the blood in animals of various ages.  
Vop. geron. i geriat. 4:247-249 '65. (MIRA 18:5)

1. Institut gerontologii AMN SSSR, Kiyev.

SHEVCHUK, V.G.

System  $\text{Li}_2\text{SO}_4 - \text{Cs}_2\text{SO}_4 - \text{H}_2\text{O}$  at  $35^\circ\text{C}$ . Zhur. neorg. khim. 10 no.9:  
2167-2169 S<sup>4</sup> 1965. (MIRA 18:10)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra khimii.



SHEVCHUK, V.G.

System  $\text{Li}_2\text{SO}_4 - \text{Cs}_2\text{SO}_4 - \text{MgSO}_4 - \text{H}_2\text{O}$  at  $35^\circ\text{C}$ .

Zhur.neorg.khim. 10 no.12:2750-2752 D '65.

(MIRA 19:1)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra  
khimii.

SHEVCHUK, V.G.

System  $\text{Cs}_2\text{SO}_4 - \text{Na}_2\text{SO}_4 - \text{H}_2\text{O}$  at  $35^\circ\text{C}$ . Zhur.neorg.khim. 10

no.12:2819-2821 D '65.

(MIRA 19:1)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra khimii.

SHVCHUK, V.G.; AVERINA, R.A.

System  $\text{Li}_2\text{SO}_4 - (\text{NH}_4)_2\text{SO}_4 - \text{MgSO}_4 - \text{H}_2\text{O}$  at  $25^\circ\text{C}$ . Zhur.neorg.khim.

10 no.12:2824-2826 D '65.

(MIRA 19:1)

1. Poltavskiy inzhenerno-stroitel'nyy institut, kafedra khimii.

SHEVCHUK, V.I.

Remarks on semiautomatic relay block systems KB TsSH. Avtom.,  
telem. i sviaz' 8 no.6:39-40 Je '64. (MIRA 17:6)

1. Nachal'nik Kulundinskoy distantzii signalizatsii i svyazi  
Zapadno-Sibirskoy dorogi.

STAL'MAKHOV, V.S.; SHEVCHUK, V.N.; ZHARKOV, Yu.D.

Analysis of the performance of a backward-wave tube in a  
cosinusoidally approximating field. Izv.vys.ucheb.zav.;  
radiofiz. 1 no.4:131-136 '58. (MIRA 12:5)

1. Saratovskiy gosudarstvennyy universitet.  
(Traveling-wave tubes)

SHEVCHUK, V. V.

Levinson, V.P., Eng Tech Sci--(M) "Study of the processes of solidifying  
metals by the solidifying method of synthesis." Ukray, 1952. 16 pp  
(Ann. of High Temp. Chem., Ukray Polytech Inst), 1000000  
(Ukr, 1952, 1000000)

-73-

66980  
SOV/81-59-13-45165

5.3200

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 13, p 63 (USSR)

AUTHORS: Shevchuk, V.U., Strom, D.A., Zeliznyy, A.M.

TITLE: The Problem of the Role of a Solid Surface at the Oxidation Pyrolysis  
of Methane<sup>1</sup>

PERIODICAL: Nauchn. zap. L'vovsk. politekhn. in-ta, 1958, Nr 50, pp 143 - 151

ABSTRACT: The dependence of the rate of oxidation pyrolysis of  $\text{CH}_4$  in a mixture of 60% Dashava gas and 40%  $\text{O}_2$  on the gas flow speed, the temperature, the partial hydrogen pressure and the wall surface value  $S$  has been investigated in a quartz pipe at 1,200 - 1,500°C and a pressure of 1 atm. It has been found that at 1,200 - 1,400°C the decomposition of  $\text{CH}_4$  to  $\text{C}_2\text{H}_2$  in the oxidation pyrolysis of  $\text{CH}_4$  proceeds with the participation of a solid surface. The quote of this participation drops with an increase in the temperature. At 1,400°C the reaction proceeds completely in the homogeneous phase.  $\text{O}_2$  introduced together with  $\text{CH}_4$  forms  $\text{CO}_2$ ,  $\text{CO}$  and  $\text{H}_2\text{O}$ , in which case the composition of the reaction products depends solely on the temperature, but not on  $S$ . Additions of  $\text{H}_2$  decrease the rate of the oxidation pyrolysis of  $\text{CH}_4$  at 1,200°C by 20%;

Card 1/2



66980

SOV/81-59-13-45165

The Problem of the Role of a Solid Surface at the Oxidation Pyrolysis of Methane

the first portions of  $H_2$  adsorbed on the walls of the vessel act more vigorously than the following ones. The authors come to the conclusion of the double role of S: at small S the homogeneous decomposition of  $CH_4$  prevails, an excessively developed S suppresses it. At the optimum value of S the rate of  $C_2H_2$  formation can be increased and the temperature of the process decreased.

V. Vasserberg

Card 2/2

BALYTA, V.I.; ZELIZNYI, A.M.; ROMANYUK, I.M.; SHEVCHUK, V.U.

Layout of equipment for the production of acetylene by the  
oxidation pyrolysis of methane. Gaz.prom. 4 no.9:36-41 S '59  
(MIRA 12:11)

(Acetylene) (Methane)

Shevchuk, V.V.

OK F.T.

25(1)  
5(1)

S/019/60/000/04/041/315  
D039/D006

AUTHORS:

Romanuk, I.M., Balyta, V.I., Berkovich, V.B., Zalizny, A.M.,  
Jesner, P.A. and Shevchuk, V.V.

TITLE:

A Tunnel Type Reactor for Obtaining Acetylene by Incomplete  
Burning of Methane in Oxygen

PERIODICAL:

Sulletan' izobreteniy, 1960, Nr 4, p 13 (USSR)

ABSTRACT:

Class 12g, J. Nr 126105 (627314/23 of 4 May 1959). The re-  
actor has an annular torch with a conical nozzle, whose annu-  
lar-space axis divides the tunnel radius into two halves. To  
intensify the process and eliminate soot formation in the re-  
action zone, the conical nozzle is fitted with blade impellers  
and a central annular duct for feeding oxygen in order to sta-  
bilize the burning.

Card 1/1

SHEVCHUK, V. U.

Effect of carbon monoxide on the reaction of the oxidative pyro-  
lysis of methane. Gaz.prom. 5 no.8:44-47 Ag '60. (MIRA 13:10)  
(Carbon monoxide) (Methane) (Pyrolysis)

80231

S/076/60/034/04/38/042  
B010/B009

11.5000  
11.1000

AUTHORS: Moin, F. B., Shevchuk, V. U.

TITLE: On the Influence of Temperature Conditions on the Velocity of Backfiring of Laminar Flames 1

PERIODICAL: Zhurnal fizicheskoy khimii, 1960, Vol. 34, No. 4, pp. 932 - 934

TEXT: The influence of the temperature of the gas mixture and tube wall upon the backfiring velocity of laminar flames of a methane-oxygen mixture was investigated. For this purpose, quartz tubes (diameter 8 mm) as well as an apparatus consisting of a vertical burner (in a heater) and various meters were used. In the case of mixtures with down to 47% by volume of  $O_2$  a distinct laminar backfiring was observed. With less than 47% by volume of  $O_2$ , backfiring was indistinct, with more than 50% by volume of  $O_2$  it was turbulent. Highly concentrated gas mixtures ( $CH_4$  51.2 and 50.0% by volume,  $O_2$  47.0 and 48.0% by volume) were used. The measured values (Figs. 1-3) show that both in highly and weakly concentrated gas mixtures the backfiring velocity increases with the heating

Card 1/2

On the Influence of Temperature Conditions on the  
Velocity of Backfiring of Laminar Flames

80231  
S/076/60/034/04/38/042  
B010/B009

temperature, while the tube wall temperature remains constant. The effect of the tube wall temperature varies, i.e., while the backfiring gradient is virtually independent in the case of high concentrations of the gas mixtures, it rises with the tube wall temperature in the case of weak concentrations of the gas mixtures. The authors agree with other authors (Ref. 3) in assuming that this difference is due to the different nature of the flash, i.e., to a thermal character in the case of weak concentrations, and a radical character in concentrated mixtures. The experiments will be continued. There are 3 figures and 3 references, 1 of which is Soviet.

SUBMITTED: August 3, 1959

Card 2/2

34338  
S/170/62/005/003/001/012  
B154/B102

11.7200  
AUTHORS:

Moin, F. B., Shevchuk, V. U.

TITLE:

Rate of passage of a laminar flame through hot methane-oxygen mixtures

PERIODICAL: Inzhenerno-fizicheskiy zhurnal, v. 5, no. 3, 1962, 10 - 14

TEXT: The effect of temperature on the velocity  $u_f$  of a flame and the rate  $w$  of laminar flame passage through methane-oxygen mixtures were investigated. The normal propagation rate of a flame was simultaneously determined for temperatures up to  $400^\circ\text{C}$ . If the value of  $\ln \frac{T_B - T_0}{T - T_0}$  is assumed to be independent of  $u_f$ , then  $u_f$  and  $w$  are connected by the following (critical) equation:

$$Pe = k \cdot S^2$$

Card 1/4

X

S/170/62/005/003/001/012  
B154/B102

Rate of passage of a laminar...

$Pe = w \cdot D/a$ ;  $S = u_f \cdot D/a$ ;  $D$ -diameter of the burner;  $a$ -thermal diffusivity of gas;  $k$ -constant;  $T_B$ -temperature of the flame front;  $T_o$ -initial temperature;  $T$ -temperature in the origin of the hot zone.  $w$  was determined by measuring the gas consumption which decreases with approaching jump. For this purpose oxygen mixtures containing  $CH_4$  between 8 and 57% and  $N$  between 0.5 and 3.3% were investigated in quartz tubes 8.5, 4.3, and 2.3 mm in diameter.  $u_f$  was determined with the burner method and direct photography. To get better results than in Ref. 4 (see below) the gas flow from the burner was formed to a flat cone. The apex angle  $\beta$  at the peak of the cone was measured and the value of  $u_f$  was calculated from  $u_f = w \sin(\beta/2)$ . The experimental results obtained for  $u_f$  show that an increase of temperature from 20 to 400°C causes an increase of  $u_f$  by 2.6 to 4.5 times. The influence of the  $CH_4$  concentration on  $u_f$  can be seen in Fig. 2. The experimental results obtained for  $w$  show that for all mixtures investigated

Card 2/4

X



S/170/62/005/003/001/012

B154/B102

Rate of passage of a laminar...

the gradient of the flame passage increases proportionally to or somewhat more slowly than  $u_f$ . If, however, the gas composition is changed at constant temperatures, then the rate of passage increases  $\sim u_f^2$ . This is explained by the fact that the thermal diffusivity is proportional to the absolute temperature and practically independent of the gas composition.

A  $Pe(S^2)$  plot of the results fits a straight line within experimental error limits. The linearity agrees well with the theoretical relation

$Pe = k \cdot S^2$ . Thus it is proved that for heated mixtures

$\ln \frac{T_B - T_o}{T - T_o}$  changes only slightly up to 400 - 500°C. There are 4 figures,

2 tables, and 7 references: 2 Soviet and 5 non-Soviet. The three references to English-language publications read as follows: Ref. 1: Lewis B., Elbe G. J. Chem. Phys., 11, 75, 1943.; Ref. 4. Sholte and Vaags. Comb. and Flame, 3, 4, 1959.; Ref. 6. Stevens T. J. Amer. Chem. Soc., 48, 1896, 1926; 50, 3244, 1928.

Card 3/4

X

Rate of passage of a laminar...

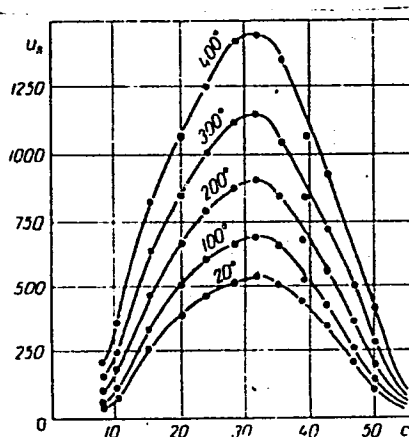
S/170/62/005/003/001/012

B154/B102

SUBMITTED: July 10, 1961

Fig. 2. Propagation rate of the flame in methane-oxygen mixtures (cm/sec) at different temperatures as a function of the  $\text{CH}_4$  concentration (%)

Fig. 2.



Card 4/4

SHEVCHUK, V.U.

Obtaining acetylene through the oxidation pyrolysis of  
methane-propane-butane mixtures. Gaz. prom. 7 no.3:36-41 '62.  
(MIRA 17:8)

KHITRIN, L. N.; MOIN, F. B.; SMIRNOV, B. B.; SHEVCHUK, V. U.

"Peculiarities of laminar and turbulent flame-backs."

report submitted to 10th Intl Symp on Combustion, Cambridge, UK, 17-21 Aug 64.

Inst Chemical Physics, AS USSR, Moscow.

KHITRIN, L.N.; MOIN, F. B.; SMIRNOV, B. B.; SHEVCHUK, V. U.

"Peculiarities of laminar and turbulent flame flashbacks."

report presented at the 10th Intl Combustion Symp, Cambridge, UK, 17-21 Aug 64.

Krzhizhanovskiy Inst of Power Engineering, Moscow.

L 20737-65 EPF(c)/EPR/EWP(j)/EWT(m) Pc-4/Pr-4/Ps-4/Pa-4 RPL/AS(mp)-2  
R/WW

ACCESSION NR: AP5001869

S/0286/64/000/023/0020/0020

AUTHORS: Shevchuk, V. U.; Fagarash, M. B.

TITLE: A method for obtaining chlorotrifluoroethylene.<sup>1</sup> Class 12, No. 166674<sup>30 B 15</sup>

SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 23, 1964, 20

TOPIC TAGS: aliphatic compound, halogen compound, hydrocarbon, chlorotrifluoroethylene, chlorodifluoroethane, dichlorofluoromethane, pyrolysis, liquid oxygen, chromatography

ABSTRACT: This Author Certificate presents a method for obtaining chlorotrifluoroethylene from halogen derivatives of aliphatic hydrocarbons. The method involves cooling and collecting the condensed products of the reaction. To enlarge the assortment of raw materials, a mixture of chlorodifluoroethane and dichlorofluoromethane is used for halogen derivatives of aliphatic hydrocarbons. This mixture is pyrolyzed at a temperature of 900-1200C under a pressure of 0.1-1 atmosphere. The products of the reaction are then collected after being condensed in liquid oxygen and separated chromatographically.

ASSOCIATION: none

SUBMITTED: 30Aug63

ENCL: 00

SUB CODE: 0C

NR REF SOV: 000

OTHER: 000

Card 1/1



L 10512-05  
ACCESSION NR: AP4038665

where  $u_l$  is the laminar burning velocity. The formula is similar to the theoretical relation as given by B. Karlovitz et al. (J. Chem. Phys., 19, 541, 1951). The main difference between the two expressions exists in their pre-exponential factors, which depend on whether the turbulence is weak, as in this case of  $u' < u_l$ , or strong, as in the case treated by Karlovitz. Orig. art. has: 3 figures and 2 formulas.

ASSOCIATION: none

SUBMITTED: 22Apr63

ENCL: 02

SUB CODE: FP

NO REF SOV: 004

OTHER: 003

2/4



ACCESSION NR: AP103866

Enclosure 01

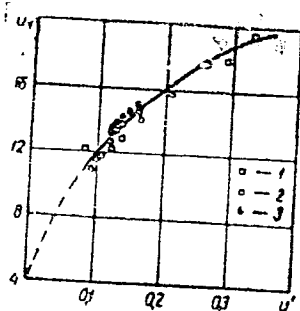


Fig. 1. Dependence of  $u_T$  (m/sec) on  $u'$  (m/sec) and the burner diameters of 6.4 (1), 8.8 (2), and 11.7 mm (3), for the mixture of 30%  $CH_4$ , 60%  $O_2$ , and 1%  $N_2$ .

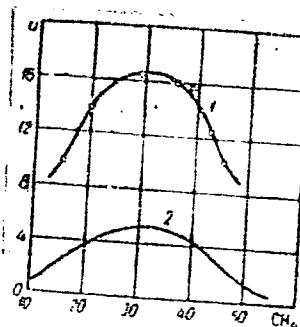


Fig. 2. The turbulent (1) and the laminar (2) burning velocity of methane-oxygen mixtures as a function of the  $CH_4$  content, for  $u' = 0.2$  m/sec.

Card 3/4

10510-65  
ACCESSION NR: AP4038665

Enclosure 02

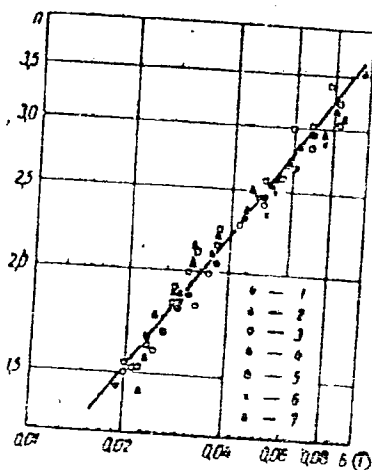


Fig. 3. Generalization of the dependence of  $u_T$  on  $u'$  and  $u_2$  for mixtures of methane and oxygen, at different concentrations of  $CH_4$  (in %): 1 - 15.2; 2 - 20.0; 3 - 30.2; 4 - 35.5; 5 - 39.0; 6 - 43.0; 7 - 45.5;

$$A = \frac{u_T}{u_2} - 1;$$

Legend 1:  $B = \frac{u'}{u_2}$

ROMANYUK, I.M.; ZELIZNYI, A.M.; SHEVCHUK, V.V.

Investigating incomplete burning in a twisted gas flow in a  
tunnel acetylene reactor. Gaz. prom. 9 no.10:34-40 '64.

(MIRA 17:12)

ZELIZNYY, A.M.; POMANYUK, I.M.; SHEVCHUK, V.U.

Increasing the productivity of a single flow reactor of oxidative  
pyrolysis of methane. Khim. prom. 40 no.12:891-894 D '64.

(MIRA 18:2)

ABADZHEV, S.S.; SHEVCHUK, V.U.

Reaction kinetics of high-temperature acetylene conversion. Gaz.  
prom. 10 no.8:33-38 '65. (MIRA 18:9)

ROMANYUK, I.M., SHEVCHUK, V U., ZELIZNYI, A.M.

Effect of the width of ignition on the process of thermooxidative  
pyrolysis of methane. Gaz. prom. 10 no.9:40-45 '65.

(MIRA 18:11)

ABADZHEV, S.S.; SIVCHUK, V.V.

Kinetics of the thermal decomposition of vinyl-acetylene and  
diacetylene. Khim. prom. 41 no. 12:891-894 D '65  
(MIRA 19:1)

✓  
ABADZHEV, S.S.; SHEVCHUK, V.U.

Determination of the kinetic parameters of reactions from  
experiments under anisothermal conditions. Zhur. fiz. khim.  
39 no.8:2055-2057 Ag '65. (MIRA 18:9)



ACC NR: AP6033275

SOURCE CODE: UR/0020/66/170/004/0893/0896

AUTHOR: Sypyak, O. I.; Moin, F. B.; Shevchuk, V. U.

ORG: none

TITLE: Study of the homogeneous stages of gas-phase reactions in a stream of inert gas

SOURCE: AN SSSR. Doklady, v. 170, no. 4, 1966, 893-896

TOPIC TAGS: <sup>nuclear component</sup> reactor, gas phase reaction, ~~wall effect~~ <sup>inert gas</sup> nuclear reactor technology,

ABSTRACT: A method and apparatus have been developed for studying gas-phase reactions, under conditions of homogeneity, i.e., excluding the effect of reactor walls. The reaction is carried out in a stream of inert gas which prevents contact of the reagents with the vessel walls. The reaction zone is located in the initial diffusion region of two concentric streams: 1) a central stream of reagents; and 2) a stream of inert gas enveloping the central stream and having the same temperature and velocity. These conditions ensure the greatest possible length for the homogeneous-reaction zone. Figure 1 shows the experimental apparatus. Section 1 is a quartz tube 300 mm in diameter and 450 mm long, equipped with an external electric heater and filled with carbon packing (grain size, 1—1.5 mm). In this section, the inert gas (nitrogen) is heated to the reaction temperature. The heated nitrogen is fed to section 2 which is 45 mm in diameter and 270 mm in length and

UDC: 541.124/.125+541.127

Card 1/3

ACC NR: AP6033275

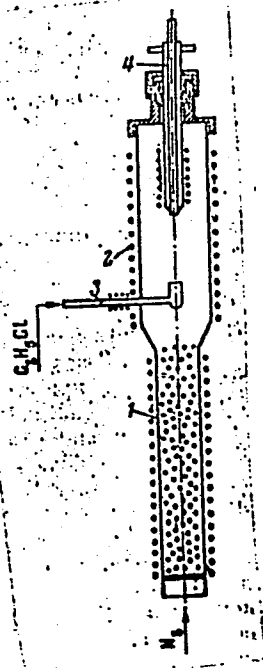


Fig. 1. Apparatus for study of homogeneous gas phase reactions

1 - Inert gas heating section;  
2 - reaction section; 3 - capillary for heating of gas of interest; 4 - sampling tube.

Card 2/3

ACC NR: AP6033275

and is equipped with an electrical heater ensuring a uniform temperature field in the entire reaction zone. The gas of interest is fed through quartz capillary 3 having an inside diameter of 0.7 mm, entering the reaction zone, and equipped with an electric heater up to the point of entry into the reactor. The heating time of the gas of interest does not exceed 0.05 sec which is a tiny fraction of the time of residence of the reagents in the reaction zone. At the point of exit of the gas of interest, the capillary is provided with a cylindrical widening, situated in the axis of the reactor, which adjusts the velocity of the gas of interest to that of the inert gas. The reaction gases are chilled and samples for analysis are taken from water-cooled quartz sampling tube 4 located at the reactor exit. To compensate for heat losses in the reaction zone which are caused by the sampling-tube cooling, this tube is equipped with an external electrical heater. The length of the diffusion zone was determined by feeding hydrogen through the capillary. The end of the diffusion zone was taken as the point where the hydrogen concentration was 0.005 vol%; gas sampling was accomplished by a capillary 2 mm in diameter which was moved along the reactor wall. Since in the method described the reaction proceeds in a zone of varying reagent concentration, the applicability of the method is limited to first-order reactions whose rate constant is independent of concentration. The method was applied to the study of the thermal-decomposition kinetics of ethyl chloride at 630—715C and a gas velocity of 15—132 cm/sec. It is expected that the new method will find use in varied kinetic studies. This paper was presented by Academician V. N. Kondratev on 19 Jan 66. Orig. art. has: 3 figures and 1 table. [WA-68]

SUB CODE: 18, 20/ SUBM DATE: 29Dec65/ ORIG REF: 003/ OTH REF: 004/  
Card 3/3

ACC NR: AP6033275

SOURCE CODE: UR/0020/66/170/004/0893/0896

AUTHOR: Sytyak, O. I.; Moin, F. B.; Shevchuk, V. U.

ORG: none

TITLE: Study of the homogeneous stages of gas-phase reactions in a stream of inert gas

SOURCE: AN SSSR. Doklady, v. 170, no. 4, 1966, 893-896

TOPIC TAGS: <sup>nuclear component</sup> reactor, gas phase reaction, ~~wall effect~~ *nuclear reactor technology*, *inert gas*

ABSTRACT: A method and apparatus have been developed for studying gas-phase reactions, under conditions of homogeneity, i.e., excluding the effect of reactor walls. The reaction is carried out in a stream of inert gas which prevents contact of the reagents with the vessel walls. The reaction zone is located in the initial diffusion region of two concentric streams: 1) a central stream of reagents; and 2) a stream of inert gas enveloping the central stream and having the same temperature and velocity. These conditions ensure the greatest possible length for the homogeneous-reaction zone. Figure 1 shows the experimental apparatus. Section 1 is a quartz tube 300 mm in diameter and 450 mm long, equipped with an external electric heater and filled with carbon packing (grain size, 1—1.5 mm). In this section, the inert gas (nitrogen) is heated to the reaction temperature. The heated nitrogen is fed to section 2 which is 45 mm in diameter and 270 mm in length and

Card 1/3

UDC: 541.124/.125+541.127

ACC NR: AP6033275

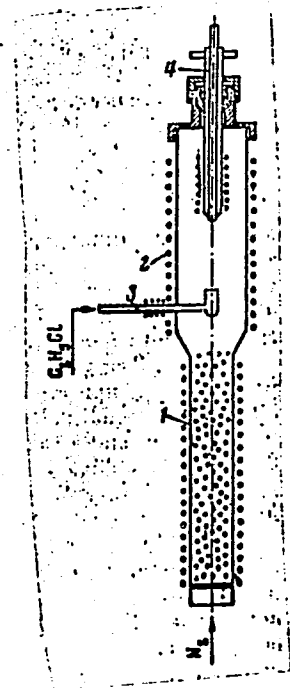


Fig. 1. Apparatus for study of homogeneous gas phase reactions

1 - Inert gas heating section;  
2 - reaction section; 3 - capillary for heating of gas of interest; 4 - sampling tube.

Card 2/3

ACC NR: AP6033275

and is equipped with an electrical heater ensuring a uniform temperature field in the entire reaction zone. The gas of interest is fed through quartz capillary 3 having an inside diameter of 0.7 mm, entering the reaction zone, and equipped with an electric heater up to the point of entry into the reactor. The heating time of the gas of interest does not exceed 0.05 sec which is a tiny fraction of the time of residence of the reagents in the reaction zone. At the point of exit of the gas of interest, the capillary is provided with a cylindrical widening, situated in the axis of the reactor, which adjusts the velocity of the gas of interest to that of the inert gas. The reaction gases are chilled and samples for analysis are taken from water-cooled quartz sampling tube 4 located at the reactor exit. To compensate for heat losses in the reaction zone which are caused by the sampling-tube cooling, this tube is equipped with an external electrical heater. The length of the diffusion zone was determined by feeding hydrogen through the capillary. The end of the diffusion zone was taken as the point where the hydrogen concentration was 0.005 vol%; gas sampling was accomplished by a capillary 2 mm in diameter which was moved along the reactor wall. Since in the method described the reaction proceeds in a zone of varying reagent concentration, the applicability of the method is limited to first-order reactions whose rate constant is independent of concentration. The method was applied to the study of the thermal-decomposition kinetics of ethyl chloride at 630—715C and a gas velocity of 15—132 cm/sec. It is expected that the new method will find use in varied kinetic studies. This paper was presented by Academician V. N. Kondratev on 19 Jan 66. Orig. art. has: 3 figures and 1 table. [WA-68]

SUB CODE: 18, 20/ SUBM DATE: 29Dec65/ ORIG REF: 003/ OTH REF: 004/  
Card 3/3

0112 2 2000, 1/2  
 MATSKIN, L.A.; KOVALENKO, K.I.; BABUKOV, V.G.; KONSTANTINOV, N.N.;  
 PONOMAREV, G.V.; PAL'CHIKOV, G.N.; PELENICHKO, L.G.; SHAMARDIN,  
 V.M.; GLADKOV, A.A.; BRILLIANT, S.G.; SHEVCHUK, V.Ya.; SOSHCHEN-  
 KO, Ye.M.; ALEKSANDROV, A.M.; BUNCHUK, V.A.; KRUPENIK, P.I.;  
 MAYEVSKIY, V.Ya.; YELSHIN, K.V.; GAK, Kh.A.; POTAPOV, G.M.;  
 KARDASH, I.M.; STEPURO, S.I.; KAPLAN, S.A.; SELIVANOV, T.I.;  
 YEREMENKO, N.Ya.; ZHUZH, A.D.; USTINOV, A.A.; GIRKIN, G.M.;  
 VOLOBUYEV, P.P.; CHERNYAK, I.L., nauchnyy red.; DESHALYT, M.G.,  
 vedushchiy red.; GENNAD'YEVA, I.M., tekhn.red.

[Combating losses of petroleum and petroleum products; materials  
 of the All-Union Conference on Means of Combating Losses of  
 Petroleum and Petroleum Products] Bor'ba s poteriami nefi i  
 nefteproduktov; po materialam Vsesoiuznogo soveshchania po bor'be  
 s poteriami nefi i nefteproduktov. Leningrad, Gos.nauchno-tekhn.  
 izd-vo nefi i gorno-toplivnoi lit-ry, 1959. 157 p. (MIRA 13:2)

1. Nauchno-tekhnicheskoye obshchestvo nefiyanoy i gazovoy pro-  
 myshlennosti.

(Petroleum industry)

L 41416-65 EWT(1)/EPA(s)-2/ EWT(m)/EPF(c)/EPF(n)-2/ENG(m)/EWA(d)/EPR/T/  
EWP(t)/EPA(bb)-2/EWP(b) Pr-4/Ps-4/Pt-10/Peb/Pu-4 IJP(c) JD/WW/JG

ACCESSION NR: AP5010469

UR/0294/65/003/002/0276/0284

AUTHOR: Labuntsov, D. A.; Shevchuk, Ye. N.; Pazyuk, P. A.

TITLE: Limiting levels of heat transfer and boiling of liquid metals

SOURCE: Teplofizika vysokikh temperatur, v. 3, no. 2, 1965, 276-284

TOPIC TAGS: heat transfer, liquid metal, sodium, potassium, mercury,  
magnesium, liquid metal boiling, heat transfer agent

ABSTRACT: Two mathematical models describing the heat transfer and boiling of liquid metals under different surface conditions are analyzed. The first model is based on the similarity in the boiling of liquid metals and common liquids, when the vapor-phase formation centers originate on the heating surface. It is shown that the highest heat-transfer level in liquid metals corresponds to vapor-phase formation conditions identical for both liquid metals and liquids. As the pressure decreases, formation of the vapor-phase on the heating surface becomes difficult, and the vapor-phase formation centers are located in the volume—the type of boiling described by the second model. In this case the number of active vapor-form-

1-3



L 41416-65

ACCESSION NR: AP5010469

mation centers on the surface which is in contact with the liquid metal (not necessarily a heating surface) is limited only to the deep scratches, depressions, and protrusions on the metal surface. The second model permits the evaluation of the lower heat-transfer levels. The rate of growth of a vapor bubble calculated for Hg and Na by the proposed equation (based on Rayleigh's dynamic model) at pressures of 1.0 and 0.02 bar and a temperature difference  $\Delta T = 50^\circ\text{C}$  compare favorably with previously published experimental data. The heat-transfer data for boiling Na-K alloy, Hg, Na, and Mg-Ti and Mg amalgams calculated by the proposed models is compared with graphs of previously published experimental data for the same liquid metal to confirm the proposed theory. The initial decrease in the heat transfer is determined by the properties of the heat-releasing surface. Heat transfer may be markedly improved by the addition of small amounts of surfactants. The boiling conditions and the heat transfer in liquid metals are sensitive to the surface microgeometry and physiochemical conditions. The problem of the thermodynamic equilibrium on the liquid-vapor interface is discussed in an appendix. Orig. art. has: 5 figures and 20 formulas. [PS]

Card 2/3

L 41416-65

ACCESSION NR: AP5010469

ASSOCIATION: Energeticheskiy institut im. G. M. Krzhizhanovskogo  
(Power Engineering Institute)

SUBMITTED: 05May64

ENCL: 00

SUB CODE: TD

NO REF SOV: 014

OTHER: 009

ATD PRESS: 3234

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Card 3/3

SHEVCHUK, Yu.

Work organization in erecting large brick blocks. Zhil. stroi. no.2:13  
'59. (MIRA 12:6)

(Building blocks)

Shelton, R., INER.

Possibilities for lowering costs of electric welding in large-  
panel housing construction. Zhil. stroi. no. 1:47 Ja '61.

(1:1 - 14:0)

(Electric welding)

SHEVCHUK, Yu., inzh.

Large-panel construction in Voroshilovsk. Zhil. stroi. no.12:22  
'61. (MIRA 15:2)  
(Voroshilovsk--Construction industry) (Apartment houses)

См. также, №. 7.

Алиев, И. И. и др. Тарыхат, №. 7. "The PKM-5 and PKM-10 weapons", Azerbaydzh. noist. shon-vo, 1948, №. 12, p. 17-18.

So: U-3261, 1 April 43, (Letopis 'Zhurnal 'ayih Statey, No. 12, 1949):

ZOLIN, Mikhail L'vovich; SHEVCHUK, Yuriy Ivanovich; AMIROV, A.D., redaktor;  
GONCHAROV, I.A., tekhnicheskii redaktor

[Pumping jack beam hangers] Kanatnye podveski dlia stankov-kachalok.  
Baku, Gos. nauchno-tekhn. izd-vo neftianoi i gorno-toplivnoi lit-ry,  
Azerbaidzhanskoe otd-nie, 1954. 34 p. [Microfilm] (MLRA 10:1)  
(Oil well pumps)

*-Shevchuk*  
MIKHAYLOV, Konstantin Fedorovich; SHEVCHUK, Yu.I., redaktor; UDALYY, A.M.,  
vedushchiy redaktor

[Deep oil well pump repair mechanic] Slesar' po remontu glubinnykh  
nasosov. Baku, Aznefteizdat, 1954. 166 p. [Microfilm] (MLRA 10:4)  
(Oil well pumps--Repairing)



ABRAMOV, M.A.; ALIVERDIZADE, K.S.; AMIROV, Ye.M.; ARENSON, R.I.; ARSEN'YEV, S.I.; BAGDASAROV, R.M.; BAGDASAROV, G.A.; BADAMYANTS, A.A.; DANIYE-  
 L'YAN, G.N.; DZHAFAROV, A.A.; KAZAK, A.S.; KERCHENSKIY, M.M.; KONYU-  
 KHOV, S.I.; KRASNOBAYEV, A.V.; KURKOVSKIY, A.I.; LALAZAROV, G.S.;  
 LARIONOV, Ye.P.; LISTENGARTEN, M.Ye.; LIVSHITS, B.L.; LISIKYAN, K.A.;  
 LOGINOVSKIY, V.I.; LYSENKOVSKIY, P.S.; MOLCHANOV, G.V.; MAY-  
 DEL'MAN, N.M.; OKHON'KO, S.K.; ROMANIKHIN, V.A.; ROSIN, I.I.; RU-  
 STAMOV, E.M.; SARKISOV, R.T.; SKRYPIK, P.I.; SOBOLEV, N.A.; TARA-  
 TUTA, R.N.; TVOROGOVA, L.M.; TER-GRIGORYAN, A.I.; USACHEV, V.I.;  
 FAYN, B.P.; CHICHEROV, L.G.; SHAPIRO, Z.L.; SHEVCHUK, Yu.I.; TSUDIK, A.A.;  
 ABUGOV, P.M., red.; MARTYNOVA, M.P., vedushchiy red.; DANIYE-  
 L'YAN, A.A.; TROFIMOV, A.V., tekhn.red.

[Oil field equipment; in six volumes] Neftianoe oborudovanie; v  
 shesti tomakh. Moskva, Gos.nauchno-tekhn.isd-vo neft. i gorno-  
 toplivnoi lit-ry. Vol.3. [Petroleum production equipment] Obo-  
 rudovanie i instrument dlia dobychi nefti. 1960. 183 p.  
 (MIRA 13:4)

(Oil fields--Equipment and supplies)

KASPIN, L.A., kand.ekonom.nauk; PAL'M, I.S., starshiy nauchnyy sotrudnik;  
KHORIKOV, A.N., starshiy nauchnyy sotrudnik; SHEVCHUK, Yu.I.,  
starshiy nauchnyy sotrudnik; AKSENOV, D.G., inzh.; EL'GORT, Ye.G.  
Prinimali uchastiye: KARAKURCHI, M.I., kand.tekhn.nauk;  
KUCHERENKO, K.R., kand.tekhn.nauk; PEDAN, M.P., nauch.sotr.; POPOV, V.Ye.,  
nauchn.sotr.; GINZBURG, S.M., inzh.; SLIN'KO, B., red.; ZELENIKOVA, Ye.,  
tekhn.red.

[Economic aspects of the construction of four- and five-story  
apartment buildings of large blocks of brick] Ekonomika vozvede-  
niya 4-5 etazhnykh zhilykh zdaniy iz krupnykh kirpichnykh blokov.  
Kiev, Gos.izd-vo lit-ry po stroit. i arkhitekt. USSR, 1960. 112 p.

(MIRA 14:4)

1. Akademiya stroitel'stva i arkhitektury USSR. Institut organi-  
zatsii i mekhanizatsii stroitel'nogo proizvodstva. 2. Sektor  
ekonomiki stroitel'nogo proizvodstva Nauchno-issledovatel'skogo  
instituta organizatsii i mekhanizatsii stroitel'nogo proizvodstva  
Akademii stroitel'stva i arkhitektury USSR (for Kaspin, Pal'm,  
Khorikov, Shevchuk, Aksenov, El'gort). 3. Nauchno-issledovatel'skiy  
institut konstruktsey (for Karakurchi, Kucherenko). 4. Glavkiyevstroy  
(for Ginzburg). 5. Nauchno-issledovatel'skiy institut stroitel'nykh  
materialov (for Pedan, Popov).

(Building, Brick)

SHEVDOV, P., inzh.

Efficient tools for finishing work. Stroitel' no.5:22 M '59.  
(MIRA 12:8)

(Building--Tools and implements)

SHEVEDEV, V. M.

Cand Tech Sci - (diss) "Study of the performance of arable aggregates on steep slopes." Moscow, 1961. 14 pp with diagrams; (Ministry of Higher and Secondary Specialist Education RSFSR, Moscow Forestry Engineering Inst); 150 copies; price not given; (KL, 7-61 sup, 249)

SHEVEKHMEN, D.; CHURAZOVA, Ye.

Use of washing machines in the glue industry. Mias.ind.SSSR 32  
no.2:44-45 '61. (MIRA 14:7)

1. Mogilevskiy kleyevoy zavod.  
(Mogilev—Glue) (Washing machines)

YUDKOVSKIY, P.A.; BULANOV, V.Ya.; ZHURAVKOV, Yu.N.; SHEVEN', A.P.

Effect of heat treatment on the strength of drills. Stan. i instr.  
34 no.12:27-28 D '63.

(MIRA 17:11)

SHEVEL', B.G.

"Experience in the Clinical Use of Radioactive Phosphorus" p. 229, in the book Experience in the Use of Radioactive Isotopes in Medicine R. Ye. KAVETSKIY and I.T. SHEVCHENKO, publishing House of the UKRAINIAN SSR, KIEV 1955, represents medical transactions of a conference held in KIEV from 18-20 January 1954.

So: 1100235

USSR/Human and Animal Physiology (Normal and Pathological) T  
The Effect of Physical Factors. Ionizing Irradiation

Abs Jour : Ref Zhur Biol., No 6, 1959, 27195  
Author : Lerner, I.P., Shevel', B.G.  
Inst : Kiev Institute for the Advanced Training of Physicians  
Title : Case of Polycythemia , Complicated by Radiation Injury  
of Vessels.  
Orig Pub : Sb. nauchn. rabot po radiol. i rentgenol. Kiyevsk. in-t  
usoversh. vrachey. Kiyev, 1957, 207-211  
Abstract : No abstract.

Card 1/1

- 172 -



BUTAKOV, D.; BOCHKOVA, V.; SHEVEL', I.; CHIZHOV, K.Ya., otv.red.; ROSHCINA,  
L., red.; TELEGINA, T., tekhn.red.

[Finances of the people's democracies] Finansy stran narodnoi  
demokratii. Moskva, Gosfinizdat, 1959. 343 p. (MIRA 13:3)

1. Nauchno-issledovatel'skiy finansovyy institut (for Butakov,  
Bochkova, Shevel').

(Finance)

BURLAKOV, M.; BUTAKOV, D.; SHKVEL', I.

General characteristics and special features of the development of the banking system in the people's democracies. Den.  
i kred. 17 no.9:34-45 S '59. (MIRA 12:12)  
(Banks and banking)

SHEVEL', I.

Agricultural credit in the people's democracies. Den. i kred. 18 no.3:  
43-55 Mr '60. (MIRA 13:2)

(Agricultural credit)

BOCHKOVA, V.; BUTAKOV, D.; BURLAKOV, M.; SHEVEL', I.; CHIZHOV, K.Ya.;  
ZABOROV, Ya., red. izd-va; POGODIN, Yu., red. izd-va; TELEGINA, T.,  
tekhn. red.

[Banks and credit in the people's democracies] Banki i kredit v stranakh narodnoi demokratii. By V.I.Bochkova i dr. Moskva, Gosfinizdat, 1961. 323 p. (MIRA 14:11)

(Communist countries—Banks and banking)

(Communist countries—Credit)

SHEVEL', I.

Credits and free help in relations among the people's democracies.  
Den. i kred. 19 no.9:61-72 S '61. (MIRA 14:9)  
(Communist countries--Foreign economic relations)  
(Communist countries--Credit)

SHEVEL', I.

Credits of the people's democracies to underdeveloped countries.  
Den. i kred. 20 no.6:39-46 Je '62. (MIRA 15:6)  
(Underdeveloped areas--Economic assistance)

SHEVEL', N.Z., uchitel'

~~Rabbit-raising~~ section of a boarding school. Biol.v shkole no.3:  
51-53 My-Je '59. (MIRA 12:9)

1. Srednyaya shkola-internat g.Novgorod-Severskiy Chernigovskoy  
oblasti.

(Rabbits)

SHEVEL', N.Z.

Experiments with rabbits. Politekh.obuch. no.11:42-43  
N '59. (MIRA 13:2)

1. Novgorod-Severskaya srednyaya shkola-internat Chernigov-  
skoy oblasti USSR.  
(Novgorod-Severskiy--Rabbit breeding)



SHEVEL', N.Z., uchitel'

Experiments with rabbits in the school rabbit farm. Biol.v shkole  
no.2:54-56 Mr-Apr '60. (MIRA 13:8)

1. Srednyaya shkola-internat goroda Novgorod-Severskogo, Chernigov-  
skoy oblasti, USSR.  
(Rabbits)

ACC NR: AR6033857 SOURCE CODE: UR/0196/66/000/008/V006/V006

AUTHOR: Shevel', S. S.; Dubrov, N. S.

TITLE: Instrument for the continuous measurement and adjustment of ultraviolet radiation intensity qm

SOURCE: Ref. zh. Elektrotehnika i energetika, Abs. 8V26

REF SOURCE: Sb. Mekhaniz. i elektrifik. s. kh. Kiyev, Urozhay, 1965, 142-143

TOPIC TAGS: uv radiation, radiation intensity, uv radiation measurement, microammeter, photoresistor

ABSTRACT: The instrument for measuring and adjusting the radiation intensity of uv sources consists of an a-c generator (100 Kc), a differential measuring bridge, an indicator, a microammeter, a relay, pickups (diodes), and two photoresistors using light filters. A 4.5 v KBS-05<sup>2</sup> battery serves as the power source. The rectified voltage is transmitted through opposite connected diodes to the photoresistor in such a manner that current difference passes through the indicator. When the conductivity of the photoresistors is equal, the indicator has no current.

Card 1/2

UDC: 535.243

ACC NR: AR6033857

However, if one of the photoresistors is irradiated, its conductivity increases and a current proportionate to the radiant flux flows through the microammeter and the relay. When irradiation increases above a given value, the relay contacts used in the corresponding switching circuit operate and source voltage decreases. The instrument can also be used for measurements in the visible region of the spectrum. Suitable light filters are used for this purpose. One illustration.  
G. L'vina. [Translation of abstract]

SUB CODE: 09/

Card 2/2

... 20-4/Pq-4/Pg-4/Pk-4/Pl-4 LJP(c) ...

... for magnetic fields report, with annual ...

... 29, no. 1, 1963, 168-171

... 1963, nuclear paragraph ...

12

GUREV, Grigoriy Abramovich; SHEVELEV, A., redaktor; IGNAT'YEVA, A.,  
tokhnicheskiiy redaktor. ~~MAN-ORIGIN~~

[How the human race originated] Kak proizoshel chelovecheskii rod.  
[Moskva] Moskovskii rabochii, 1955. 102 p. (MLRA 8:11)  
(Man--Origin)

SHEVELEV, A.

Some aspects of technical standardization in the simultaneous  
servicing of several machines. Sots.trud 4 no.9:74-79

S '59. (MIRA 13:1)  
(Machinery industry--Production standards)

SHEVELEV, A., general-mayor tekhnicheskikh voyak.

Technical inspection of automobiles. Voen.vest. 36 no.4:37-39  
Ap '56. (MLRA 9:8)  
(Automobiles, Military--Inspection)

*SHEVELEV, A.*  
SHEVELEV, A., inzh.

Economic indices in constructing a large-block apartment house  
in Rostov-on-Don. Gor.i sel.stroi. no.8/9:23-24 Ag-S '57.

(MIRA 10:12)

(Rostov-on-Don--Apartment houses) (Construction industry--Costs)



SHEVELEV, A., inzh.

Determining the technical readiness of a structure by standard  
labor expenditures, Stroitel' no.1:14 Ja '58. (MIRA 11:2)  
(Building--Estimates)

SHEVELEV, A., inzh.

Determining the technical readiness of constructed buildings according  
to labor expenditures. Na stroi. Mosk. 1 no.8:24-25 Ag '58.

(MIRA 11:10)

(Building)

100-100000-10  
OBUKHOV, V., inzh.; SHEVELEV, A., inzh.

Adobe construction. Sel'. stroi. 12 no.2:5-8 F '58. (MIRA 11:2)  
(Moscow Province—Building, Adobe)

SHEVELEV, A.

Improve the planning and accounting for labor productivity in  
rural construction. Sots. trud 6 no. 1:84-88 Ja '61.  
(MIRA 14:1)

(Farm buildings)

(Construction industry--Labor productivity)

SHEVELEV, A.A.

104-3-19/45

AUTHOR: Modylevskiy, D.N. and Shevelev, A.A., Engineers.

TITLE: The 30th Anniversary of the Shterovsk State Regional Power Station named after F.E. Dzerzhinskiy. (30-letiyе Shterovskoy gosudarstvennoy rayonnoy elektrostantsii imeni F.E. Dzerzhinskogo)

PERIODICAL: "Elektricheskiye Stantsii" (Power Stations), 1957, Vol. 28, No.3, p. 60 (U.S.S.R.)

ABSTRACT: This note gives a brief account of the history of the station which was started in October, 1926, with a 20 MW set and which by 1931 was a station of 152 MW. The station was destroyed during the war. By 1944, three sets totalling 54 MW were running, by 1946 a further 10 MW had been added and since then the sets have been super-posed by back-pressure sets using steam conditions of 126 atm and 500 C. The successes of the station in respect of fuel economy and other matters are described.

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PAVLOVSKIY, G.I., kand. tekhn. nauk.; SHEVELEV, A.A., inzh.

Temperature fields in flange couplings of turbines during initial heating. *Energomashinostroenie* 4 no.9:14-17 S '58. (MIRA 11:11)  
(Steam turbines)

91-58-7-6/27

AUTHORS: Dutikov, S.S.; Shevelev, A.A.; Vaytsman, V.M., Engineers  
and Vnukov, A.K., Candidate of Technical Sciences

TITLE: Exchange of Experience (Obmen opytom). The Automated Operation of Mills (Avtomatizatsiya raboty mel'nits).

PERIODICAL: Energetik, 1958, Nr 7, pp 19-20 (USSR).

ABSTRACT: In 1957, 5 drum ball mills (4 mills of "Sh-16" type and 1 biconical mill of "ShK-25" type) were automated according to the design suggested by Yuzhnoye otdeleniye ORGRES (the "ORGRES" South Branch Office). The following equipment was utilized: electronic controllers of "ER-III" type on 2 mills and electromechanical direct feedback columns of the "Energodetal' " plant on 3 mills. Their structural details and operation are described. The first experimental service of this automated system proved its operational stability and wide control range, as well as easy maintenance. The various requirements to be met for automating mills, such as good dust system, continuous aeration etc. are

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Exchange of Experience

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outlined. The "dust level" automation of mills must guarantee a decrease in consumption of electric energy for the preparation of pulverized coal by 3 to 4 kwh per ton of milling. There are 2 diagrams.

1. Ball mills--Operation 2. Ball mills--Electronic controls

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S/143/60/000/007/005/010  
A189/A029

AUTHOR: Pavlovskiy, G.I., Candidate of Technical Sciences, Shevelev,  
A.A., Engineer

TITLE: Heating of<sup>n</sup> Bodies With Regard to Thermal Stresses

PERIODICAL: Energetika, 1960, Vol 3, Nr 7, pp 81-87

TEXT: The heating and cooling<sup>n</sup> process of bodies having a regular geometric shape is analyzed and the arising thermal stresses are calculated. General relations between the thermal stresses and the difference of temperatures are derived. The derived relations are applicable to calculations of thermal stresses in bodies of any shape with a sufficient accuracy for practical use. A method is given for determining the optimum cooling conditions of bodies whose permissible thermal stresses are known. There are 4 graphs and 3 Soviet references.

ASSOCIATION: Khar'kovskiy politekhnicheskii institut imeni V.I. Lenina  
(Khar'kov Polytechnical Institute imeni V.I. Lenin);

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S/143/60/000/007/005/010  
A189/A029

Heating of Bodies With Regard to Thermal Stresses

Kafedra obshchey teplotekhniki (Department of General Heat  
Engineering)

SUBMITTED: July 6, 1959

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S/143/60/000/009/005/006  
A189/A026

Additional Heating-Up as a Method of Eliminating Thermal Stresses in a Turbine Cylinder

ASSOCIATION: Khar'kovskiy politekhnicheskii institut imeni V.I. Lenina (Khar'kov Polytechnical Institute imeni V.I. Lenin)

PRESENTED: Kafedra teplotekhniki (Department of Heat Engineering)

SUBMITTED: February 26, 1960

Card 2/2

WILSON, R. L.; SHEVLEY, R. L.

Temperature stresses in the flanged joints of steam turbines.

Gray's R&E 29 no. 2:187-193 '60.

(SER. 24:10)

(Steam turbines)

(Thermal stresses)